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High-Sensitivity Observation of Dipolar Exchange and NOEs between Exchangeable Protons in Proteins by 3D Solid-State NMR Spectroscopy

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While NMR structures of proteins are nearly always for molecules in solution, the vast majority of protein structures have been obtained by X-ray diffraction on frozen crystals. Problems posed by membrane proteins, insoluble protein aggregates, and differences between some X-ray and solution NMR structures all provide compelling reasons for developing robust solid-state NMR (ssNMR) methods for determination of protein structure. A great deal of progress has been made in this regard in the past decade. Approaches for obtaining full ¹³C and ¹⁵N assignments¹⁻⁴ and for assigning backbone amide ¹H shifts⁵ have been demonstrated on small proteins or peptides. A wide range of techniques have been developed to measure ¹³C–¹³C, ¹³C–¹⁵N, and ¹⁵N–¹⁵N distances.^{6–8} Such methods, in combination with patterned isotopic enrichment, produced the first ssNMR structure of a small crystalline protein domain in the past year.⁹

These pioneering advances, relying on heteronuclei, have been impressive technical accomplishments. However, if ssNMR is to approach solution methods in scope and capability, techniques that exploit the higher sensitivity and larger distance measuring range of ¹H's must be developed. The experimental realization of ¹H detection for high-resolution magic angle spinning (MAS) ¹H/¹⁵N heteronuclear single quantum coherence (HSQC) ssNMR spectroscopy makes deployment of 3D methods and the measurement of tertiary contacts between ¹Hs very much more feasible.^{10–12} In this Communication, we report a new 3D ssNMR technique analogous to the ¹H-detected 3D ¹H/¹⁵N HSQC NOESY method. We demonstrate that this sensitive 3D spectroscopy can confirm assignments of backbone amide resonances and observe long-range contacts in application to a β -turn in human ubiquitin.

Figure 1 diagrams the basic pulse sequence developed. This is intended for application to exchangeable ¹H's in ¹⁵N- and ²Henriched protein samples, where the combination of magnetic dilution and fast MAS has been shown by us12 to provide average ¹H line widths of only 0.22 ppm. ¹H evolution in t_1 under WALTZ-16 ¹⁵N decoupling and MAS at \sim 20 kHz is projected onto the spinlock field in the first cross-polarization (CP) step. Magnetization is transferred to ¹⁵N, which evolves for t_2 under ¹H decoupling, and the desired frequency-labeled component is projected onto the z-axis by a $\pi/2$ pulse. While this ¹⁵N magnetization is stored along z, the ¹H decoupling is continued so that the experiment operates in a constant power mode regardless of the length of t_2 . This is key to reducing t_2 noise as well as being a central ingredient in efficient water suppression. A further delay of t_w of ~10 ms provides for dephasing of unwanted transverse magnetization. ¹H magnetization that is frequency labeled by ¹H chemical shift in t_1 and ¹⁵N chemical shift in t_2 is then created with a second CP step and immediately placed along the z-axis. Spin exchange of several



Figure 1. Pulse sequence for 3D ssHSQC-detected dipolar exchange spectroscopy. Solid bars represent $\pi/2$ pulses. The ¹H $\pi/2$ pulse length is 2.4 μ s, and the ¹⁵N $\pi/2$ pulse length is 4.6 μ s. TPPM decoupling was performed at 100 kHz. CP transfers use a ¹⁵N rf field of ~55 kHz and a ¹H rf field of ~75 kHz with a spin rate of ~20 kHz. $t_{2max} = 30$ ms, and $t_{1max} = 5$ ms. The phase ϕ is switched between -y and x phases to acquire quadrature data in t_2 . The ¹⁵N RF level is set to ~2.5 kHz during WALTZ-16 decoupling. The ¹H carrier frequency is placed on the water resonance. The ¹⁵N $\pi/2$ pulse, prior to final CP and receiver phase, is switched between y and -y on alternate scans. For a description of the 800-MHz instrument used and other experimental conditions, see ref 12.

types can then occur during the period τ_{mix} . We have used RFDR¹³ as a ¹H⁻¹H dipolar recoupling method in this work, but any number of other longitudinal or transverse techniques can be applied.^{6,14} After the appropriate mixing time, the magnetization is placed in the *xy* plane, and a high-resolution ¹H MAS spectrum acquired.

For magnetization originating on the *k*th amide 1 H, the real component of the 3D time domain signal is of the form

$$S_{k}(t_{1},t_{2},t_{3}) = \sum_{m} a_{k} \cos(\omega_{\mathrm{H}_{k}}t_{1}) \cos(\omega_{\mathrm{N}_{k}}t_{2}) \cos(\omega_{\mathrm{H}_{k}}t_{3}) + c_{km} \cos(\omega_{\mathrm{H}_{k}}t_{1}) \cos(\omega_{\mathrm{N}_{m}}t_{2}) \cos(\omega_{\mathrm{H}_{m}}t_{3})$$

where the sum extends over the m amides that exchange or crossrelax with k during τ_{mix} . Each ω_1 plane has an autopeak of intensity a_k and cross-peaks for each of the *m* exchange partners with amplitudes c_{km} . Autopeaks fall on the line $\omega_1 = \omega_3$ in any ω_1 plane. Walks through the protein start by picking an autopeak, moving along ω_2 to find cross-peaks at frequencies ω_3' , and continuing by confirmation of an autopeak in the corresponding plane at $\omega_1' =$ ω_3' . An example is provided in Figure 2, showing strip plots for the β -turn in ubiquitin comprising residues F45-K48. This particular data set used RFDR for 1.2 ms as the dipolar recoupling during τ_{mix} . Other experimental details are listed in the figure caption, but it should be noted that this very high signal-to-noise result was acquired on just 500 nmol of protein in only 48 h. We estimate the 200 largest of the over 300 peaks picked by computer search would easily be identified with half the signal-to-noise, or acquired in one-fourth the time. The autopeaks for A46 and G47 in ubiquitin are outliers and are assigned on the basis of similarity to the solution NMR shifts.15

K48 and F45 can be assigned on the basis of this walk. Other short segments in the protein can be also be walked just as readily.

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Figure 2. Strip plots of $\omega_2\omega_3$ planes at ω_1 shift indicated at top. Each strip is 1 ppm wide in ω_2 and centered at the ¹⁵N shift indicated. Signal-to-noise ratio is indicated by the slice taken through the leftmost strip. Forty-eight t_1 points were acquired using a dwell time of 104 μ s in t_1 , and 96 complex points were acquired in t_2 using a dwell time of 312 μ s. ¹H carrier was placed on the water resonance. Sixteen scans per point, 1 s recycle delay, ¹H resonance frequency 799.55 MHz. The walk from A46 back to F45 shown in red. The ubiquitin sample was prepared using 2-methyl-2,4-pentanediol- d_{12} as precipitant, as described in ref 12.



Figure 3. Depiction of the β -turn comprised of F45–K48 in ubiquitin, showing the interspin distances involved. The water molecules closest to A46 and G47 are also shown.

Cross-peaks corresponding to all possible ${}^{1}\text{H}{-}{}^{1}\text{H}$ contacts among these four spins are observed. To the left of Figure 2 is a slice taken along the ω_3 axis for F45, indicating the relative intensities of the autopeaks and cross-peaks. The strongest cross-peak for F45 is across the turn to K48, while the weakest is the sequential crosspeak with A46. This is the expected result if the β -turn structure in this crystal form¹⁶ is similar to that in the reported X-ray structure.¹⁷ Adding hydrogens to the crystal structure,¹⁸ the ¹H contacts involved in this turn are calculated to cover the range of 2.7–4.3 Å. Some of these are indicated in the structural diagram in Figure 3. As the cross-peak intensities would indicate, the K48 and F45 amide protons are quite close, being separated by only 2.74 Å, while the corresponding A46–F45 distance is much longer, at 4.11 Å. Full utilization of the many contacts observed of this type will require a complete independent assignment of the ¹⁵N and ¹H shifts, and this work is in progress.

In addition to amide-to-amide cross-peaks, many amides have cross-peaks with bound water. The kinetics of the magnetization exchange of individual amides with water can readily be studied by setting t_1 to zero in the pulse sequence of Figure 1, and collecting 2D spectra as a function of τ_{mix} . An even simpler measurement is to also set t_2 to zero and follow the τ_{mix} time course of the amide and crystal water bands in 1D spectra. In this manner, an average cross-relaxation time constant of ~90 s⁻¹ is observed, independent of whether dipolar recoupling is used. However, if the exchange is monitored in the rotating frame by extending the final *y*-phase ¹H CP pulse into τ_{mix} and eliminating the pair of flip up and down ¹H pulses, the kinetics are quite different. These observations are consistent with the amide-to-water magnetization exchange being NOE relaxation in the solid state, as opposed to chemical exchange¹⁹ or spin flips due to static dipolar interactions.

In summary, a sensitive solid-state analogue of the 3D 1 H/ 15 N HSQC-detected NOESY experiment has been described, and longrange nonsequential 1 H $^{-1}$ H contacts have been observed. The method provides an entry to structure determination in nanocrystalline 15 N/ 2 H-enriched protein samples and is expected to have its greatest utility in study of the physical chemistry and dynamics of proteins. In this vein, the observation of numerous NOEs between amide protons and structural water highlights the type of information we expect this new approach will provide in the study of macromolecular systems.

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References

- McDermott, A.; Polenova, T.; Bockmann, A.; Zilm, K. W.; Paulson, E. K.; Martin, R. W.; Montelione, G. T. *J. Biomol. NMR* 2000, *16*, 209– 219.
- (2) Straus, S. K.; Bremi, T.; Ernst, R. R. J. Biomol. NMR 1998, 12, 39–50.
 (3) Rienstra, C. M.; Hohwy, M.; Hong, M.; Griffin, R. G. J. Am. Chem. Soc.
- 2000, 122, 10979-10990.
 (4) Pauli, J.; Baldus, M.; van Rossum, B.; de Groot, H.; Oschkinat, H. *ChemBioChem* 2001, 2, 272-281.
- (5) van Rossum, B. J.; Castellani, F.; Pauli, J.; Rehbein, K.; Hollander, J.; de Groot, H. J. M.; Oschkinat, H. J. Biomol. NMR 2003, 25, 217–223.
- (6) Baldus, M. Prog. Nucl. Magn. Reson. Spectrosc. 2002, 41, 1-47.
- (7) Hohwy, M.; Rienstra, C. M.; Griffin, R. G. J. Chem. Phys. 2002, 117, 4973–4987.
- (8) Jaroniec, C. P.; Filip, C.; Griffin, R. G. J. Am. Chem. Soc. 2002, 124, 10728-10742.
- (9) Castellani, F.; van Rossum, B.; Diehl, A.; Schubert, M.; Rehbein, K.; Oschkinat, H. Nature 2002, 420, 98–102.
- (10) Reif, B.; Griffin, R. G. J. Magn. Reson. 2003, 160, 78-83.
- (11) Chevelkov, V.; van Rossum, B. J.; Castellani, F.; Rehbein, K.; Diehl, A.; Hohwy, M.; Steuernagel, S.; Engelke, F.; Oschkinat, H.; Reif, B. J. Am. Chem. Soc. 2003, 125, 7788–7789.
- Chem. Soc. 2003, 125, 7788-7789.
 (12) Paulson, E. K.; Morcombe, C. R.; Gaponenko, V.; Dancheck, B.; Byrd, R. A.; Zilm, K. W. J. Am. Chem. Soc. 2003, 125, in press.
- (13) Bennett, A. E.; Ok, J. H.; Griffin, R. G.; Vega, S. J. Chem. Phys. 1992, 96, 8624–8627.
- (14) Karlsson, T.; Popham, J. M.; Long, J. R.; Oyler, N.; Drobny, G. P. J. Am. Chem. Soc. 2003, 125, 7394-7407.
- (15) Wang, A. C.; Grzesiek, S.; Tschudin, R.; Lodi, P. J.; Bax, A. J. Biomol. NMR 1995, 5, 376–382.
- (16) Igumenova, T. I. Assignment of uniformly carbon-13-enriched proteins and optimization of their carbon line shapes; Columbia University: New York, 2003; p 199.
- (17) Vijaykumar, S.; Bugg, C. E.; Cook, W. J. J. Mol. Biol. 1987, 194, 531–544.
- (18) Word, J. M.; Lovell, S. C.; Richardson, J. S.; Richardson, D. C. J. Mol. Biol. 1999, 285, 1735–1747.
 (19) Grzesiek, S.; Bax, A. J. Biomol. NMR 1993, 3, 627–638.

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